STEREOSELECTIVE SYNTHESIS OF DL-SENEPOXYDE<sup>1)</sup>

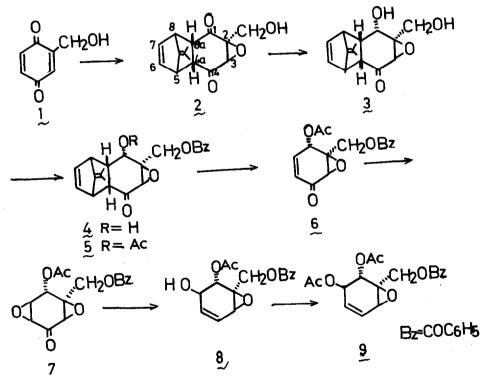
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Senepoxyde<sup>2)</sup> (9) which was isolated from medicinal plant, <u>Uvaria Catcatpa</u> in France, is known as one of naturally occurring highly oxygenated cyclohexane derivatives<sup>3)</sup>. In this communication, we would like to describe the total synthesis of <u>dl</u>-senepoxyde in stereoselective manner.

Diels-Alder reaction of 2-hydroxymethyl-1,4-benzoquinone 1 with dimethylfulvene and subsequent epoxidation of the adduct gave 2, m.p.  $101 \sim 102^{\circ}$ ,  $C_{15}H_{16}O_4$ ,  $)_{max}^{KBr}$  1710 cm<sup>-1</sup>;  $\int_{TMS}^{CDCh_3} 1.53$  (6H, s,  $=^{CH_3}$ ), 3.42 (2H, m, -CHCO-), 3.56 (1H, s,  $\stackrel{O}{\longrightarrow}$  H), 3.72 (2H, m,  $=^{\sim}$  H), 3.85, 4.00 (2H, ABq,  $j_{AB}$  = 14 Hz -CH<sub>2</sub>-O), 6.19 (2H, m,  $\stackrel{H}{=}$ ). The fulvene part of the compound 2 plays an important role not only to produce the regio- and stereoselectivity but also to stabilize the intermediates in the course of the synthesis and could be easily removed in later stage. The stereochemistry of compound 2 was determined by the fact that two protons attached to ring juncture C-4a and C-8a in <u>endo</u> compound 2 appeared at  $\int 3.42$ , lower field compared with the signal at  $\int 2.80$  which is shielded by double bond in <u>exo</u> compound<sup>4,5)</sup> and epoxidation should be preferably occurred from <u>exo</u> side to give a sole product 2. Regioselective reduction of 2, with 1.1 equivalent of sodium borohydride utilizing neighboring hydroxyl group effect gave a diol 3, in 73.5% yield, m.p. 149.5~150°, C<sub>15</sub>H<sub>18</sub>O<sub>4</sub>,  $\bigvee_{max}^{KBT}$  3500, 3300, 1705 cm<sup>-1</sup>;  $\begin{pmatrix} \text{acctone-d6}_{1.48} \\ \text{(6H, s, } (H, s, (H, A)) = 2.74 (1H, A) part of ABXY, JAB = 12 Hz, JAX = 3Hz, JAY = 5 Hz$ ,  $\stackrel{H}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{H}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{H}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{H}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{H}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{H}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{H}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{H}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{H}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{H}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{O}{}$ ,  $\stackrel{H}{}$ ,  $\stackrel$ 



Treatment of the diol 3 with benzoyl chloride afforded a benzoate 4, m.p. 150.5°, C<sub>22</sub>H<sub>22</sub>O<sub>5</sub>, which was further acetylated with acetic anhydride-pyridine to an acetate 5, m.p. 152°, C<sub>2</sub>, H<sub>2</sub>, O<sub>6</sub>, <u>m/e</u> 408 (M<sup>+</sup>); )  $\operatorname{KBr}_{max}$  1730, 1715, 1600 cm<sup>-1</sup>;

 $\int \frac{\text{CDCl}_3}{\text{TMS}^3} 1.47$  (6H, s,  $= \int \frac{\text{CH}_3}{3}$ ), 2.00 (3H, s, OCOCH<sub>3</sub>), 2.97 (1H, A part of ABXY,  $\underline{AB} = 11 \text{ Hz}$ ,  $\underline{AX} = 3 \text{ Hz}$ ,  $\underline{AY} = 5 \text{ Hz}$ ,  $\underline{AC}$  ), 3.15 (1H, B part of ABX', JAB = 11, JBX = 3.5 Hz  $\xrightarrow{H}$  ), 4.06, 4.73 (2H, ABq, JAB = 12 Hz, -CH<sub>2</sub>-0-), 5.89 (1H, dd, J = 5 Hz, J = 3 Hz,  $= 10^{-1}$  ), 5.99 (1H, d, J = 5 Hz, -CHOAc), 6.34 (1H, dd, J = 5 Hz, J = 3 Hz,  $-\frac{H}{2}$ ), 7.50 (3H, m, ArH), 8.04 (2H, dd,  $\underline{J} = 9 \text{ Hz}$ ,  $\underline{J} = 2 \text{ Hz}$ , ArH). Retro Diels-Alder reaction<sup>5)</sup> of the acetate 5 in diglyme by heating at 160°C for lhr under atmosphere of mitrogen gave an epoxycyclohexenone 6,  $\underline{m/e}$  302 (M<sup>+</sup>), ) meat 1750, 1720, 1690, 1600 cm<sup>-1</sup>;  $\int \frac{\text{CDCl}}{\text{TMS}} 2.05 \ (3\text{H}, \text{s}, -0\text{COCH}_3), \ 3.57 \ (1\text{H}, \text{dd}, \underline{J} = 1.5, \underline{J} = 1.5 \text{ Hz}, \ \underline{\swarrow}^{0} \ H$ ), 4.25, 5.00 (2H, ABq, J = 12 Hz, -CH<sub>2</sub>-O-), 5.98 (1H, m, -CHOAc), 6.11 (1H, m,  $H_{0}$  ), 6.67 (1H, dd, J = 10 Hz, J = 5 Hz,  $H_{2}$  ) in 78% yield. Bpoxidation and subsequent acetylation yielded a diepoxide  $\zeta$ ,  $\gamma_{max}^{neat}$  1745, 1730, 1600 cm<sup>-1</sup>;  $\begin{cases} CDC_{13} \\ TMS \end{cases}$  2.10 (3H, s, -OCOCH<sub>3</sub>), 3.47 (2H, m,  $\swarrow^{0}$  H ), 3.59 (1H, dd, J = 4 Hz, J = 2 Hz, 2 Hz, 2 Hz, 2 Hz, 4.00, 4.79 (2H, ABq, JAB = 13 Hz, -CH<sub>2</sub>-O), 6.00(1H, s, -CH-OAc), 7.50 (3H, m, ArH), 8.05 (2H, dd, J = 8 Hz, J = 2 Hz, ArH). In the NMR spectrum of the diepoxide 7, the fact that a signal at  $\delta$  6.00 due to  $\alpha$  -proton to acetoxyl group has rather small half width (J = 3 Hz) indicates that the hydrogen occupies trans position with a dihedral angle of nearly  $90^{\circ}$ to the vicinal hydrogen and the stereochemistry must be as shown. Finally reductive cleavage of an epoxy ring in the diepoxide with hydrazine hydrate (ca. 2 equivalent, in benzene-isopropyl alcohol-THF) proceeded regioselectively to give 8 which was directly acetylated with acetic anhydride-pyridine to afford dl-senepoxyde m.p. 97~98°, C18 H18 07. The synthetic senepoxyde is identical with natural specimen in IR, NMR, MS spectra, and behavior on TLC.

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## References and footnotes

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